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Determination of the
Molarity of a Sulfuric
Acid Solution strans
Trial 1 Trial 2 Trial 3

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Trial 4* Starting Buret

Reading: 1.74 L

On 15 L 0-59 L Ending

Buret Reading: 11.83

L 10-10 L 10.59 L 10.59 mL of

Titrant Used:

10.09 L / 9.95 L 10.00 L

mmole of NaOH:

1.009 mol / 9.95 mol 1.0

ml mmole of H₂SO₄

0.5 mol / 5.0 mol 0.5 5.0 mol

Molarity of H₂SO₄

Solution 0.05 M 0.5

M 0.05 M 0.05 M

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Reaction 1: 1 NaHCO_3

$(s) + 1 \text{ NaOH} (s) + 1$

$\text{CO}_2 (g)$ Reaction 2:

$2 \text{ NaHCO}_3 (s) \rightarrow 1 \text{ Na}_2$

$\text{CO}_3 (s) + 1 \text{ CO}_2 (g)$

$+ 1 \text{ H}_2\text{O} (g)$

Reaction 3: 2 NaHCO_3

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3(s) 1 Na₂O(s) + 2.

Propose a voltaic cell that would coat a piece of copper with platinum. Include the half reactions for your cell and the voltage potential based on the.

CHEM 103 - UIUC

The Chemistry 103
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Fall 2016 and

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carbonless laboratory notebook can be purchased (Wiscard only) in room 1375 (chemistry computer lab) during the first two weeks of classes and later through the first-floor stockroom. Lab notebooks (with carbonless copies) purchased from the University Bookstore are also acceptable.

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Chemistry (4th
Edition) Burdge, Julia
Publisher McGraw-Hill
Publishing Company
ISBN

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Approach through an

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in the stockroom 111.

Band-aids for minor ...

The answer is your

local wastewater

treatment facility,

which operates 24/7

to make sure

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1. Identify the group

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for each unknown.

Your unknowns will be from three different groups. (Example: A K^+ salt, a halogen, and an acid.) 2.

Identify the specific compound from each group. (Example: KCl , Br_2 and HCl) 3.

Explain the reasoning that enabled you to come to these conclusions.

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One of the most
important operations
in a chemistry
laboratory is the
massing of objects.
Since chemistry is an
exact science, the

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massing of
substances which
enter or result from a
chemical change
must be done with the
best possible ...

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