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Properties of Aqueous

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Reactions The Ultimate

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Note: Not all of the
reactions will occur. For

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those that do not, write no
reaction. 1. A piece of
copper is dropped into a
container of water. No
reaction 2. Liquid bromine
is added to a container of
sodium iodide crystals. Br
 $(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) +$

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12(s) 3. An aluminum strip is immersed in a solution of silver nitrate.

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tin(II) carbonate sodium
hydrogen carbonate NaHCO₃ 1.
2. 3. 4. 5. 6. 8. 9. 10. 11.
vanadium(V) oxide H₂O
dihydrogen monoxide (NH₃) CO
ammonium oxalate 4 224
polonium(VI) thiocyanate
Po(SCN)₆ tetrphosphorus

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decaoxide P O 4 10 12. 13.

15. 16. 17. 18. 19.

manganese(VII) oxide

copper(II) dihydrogen

phosphate

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ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions.

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many of the transuranium
elements were prepared at
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Center, University of
California at Berkeley,

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California, U.S.A., it is
logical to find the
following names:

Americium—named after the
Americas (after all, one
must be patriotic!).

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Jr and Jane D Smith question
requiring the determination
of products of chemical
reactions (the Equations
section) five of these eight
equations to answer (Note:
Additional responses will
not be graded or counted If

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Answers Chapter 7 Note: Not
all of the reactions will
occur. For those that do
not, write no reaction. 1.

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